Biodegradation of polyethylene and polypropylene

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Polyethylene and polypropylene are the two polyolefins with wide ranging applications. They are recalcitrant and hence remain inert to degradation and deterioration leading to their accumulation in the environment, and, therefore creating serious environmental problems. In this review, biodegradation of these two polymers under in vitro conditions is reported. An attempt has been made to cover the mechanism of biodegradation, the various bacterial and fungal organisms that have been reported for the same, methods adopted for the studies and different characterization techniques followed to measure the extent of degradation

Keywords: polyethylene, polypropylene, biodegradation, in vitro

Introduction

The myriad applications of polymers in almost all the fields ranging from sophisticated articles such as, prosthetic hips and knee joints to disposable food utensils implies their significance and importance in our day to day life. Thus, enormous production and utilisation of polymers lead to their accumulation in the environment. Since not easily degraded by microorganisms, today they have become a serious source of pollution affecting both flora and fauna.

Polyolefins or saturated polymers have a broad range of applications. Polypropylene (PP) and polyethylene (PE), expressed as C_EH_{2E}, are most widely used linear hydrocarbon polymers. The versatility of these polymers arises from the fact that they are made from cheap petrochemical feed stocks through efficient catalytic polymerisation process and their ease of processing to various products. The range of their applications include, food packaging, textiles, lab equipments, and automotive components. PP has a methyl group instead of one of the hydrogens present in PE, on every other carbon, which gives rise to the existence of three stereoisomeric forms namely, atactic, isotactic, and syndiotactic1. This stereoregular polymer was first synthesised by Ziegler and Natta with propylene as the monomer. Metallocene catalysts can also be used for its synthesis. Industrially applicable PE was first synthesised in 1933 by Eric Fawcett and Reginald Gibson at ICI chemicals². PE is totally linear and available with varying range of densities from 0.91 to 0.97 g/cm³. Low density PE has branching at random places leading to low packing of the polymer chains, whereas the high density PE is more linear with minimal branching leading to high packing density¹.

As reported by American Plastic Association, percentage distribution of PP, high density polyethylene (HDPE), linear low density polyethylene (LLDPE) and low density polyethylene (LDPE) are 18.4%, 17.4%, 12.1% and 8.2%, respectively in terms of sales and use in the year 2004 in the United States, Canada, and Mexico'. Non-degradable plastics accumulate in the environment at a rate of 25 million tonnes per year*. Extensive use of non-biodegradable thermoplastics and the rate at which they accumulate in the environment, makes the humankind to realise the necessity to find its environmental impact. As the polymer usage is unavoidable, ways have to be found to (1) Enhance the biodegradability of the polymers by blending them with biodegradable natural polymers such as starch⁵⁻¹⁹ or cellulose²⁰ etc; (2) Mixing with prooxidants^{5,21,22} so that they are easily degraded and (3) Isolate23 and improve microorganisms that can efficiently degrade these polymers. In order to attempt the third option the mechanism of biodegradation should be understood.

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Overview of Biodegradation of Polymers

A general overview of biodegradation of polymers over a period of time is schematically represented in Fig. 1. Polymeric materials released into the environment can undergo physical, chemical and biological degradation or combination of all these due to the presence of moisture, air, temperature, light (photo-degradation), high energy radiation (UV, γ-radiation) or microorganisms (bacteria or fungi). The rates of chemical and physical degradation are higher when compared to that of biodegradation. Also, physical and chemical degradation facilitates microbial degradation and complete mineralisation of the polymer happens due to biodegradation, which is generally the final step^{24,25}.

Mechanism of Biodegradation

Biodegradation of polymers involves following steps:

- Attachment of microorganism to the surface of the polymer
- Growth of microorganism utilising the polymer as the carbon source
- 3. Primary degradation of the polymer and
- 4. Ultimate degradation

Microorganisms can attach to the surface, if the polymer surface is hydrophilic. Since PP and PE have only CH2 groups, the surfaces are hydrophobic. Initial physical or chemical degradation leads to the insertion of hydrophilic groups on the polymer surface making it more hydrophilic (insertion of hydrophilic groups also decreases the surface energy). Once the organism gets attached to the surface, it start growing by using the polymer as the carbon source. In the primary degradation, the main chain cleaves, leading to the formation of low-molecular weight fragments (oligomers), dimers or monomers24. The degradation is due to the extra cellular enzymes secreted by the organism. These low molecular weight compounds are further utilised by the microbes as carbon and energy sources. Small oligomers may also diffuse into the organism and get assimilated. The ultimate products of degradation are CO2, H2O and biomass under aerobic conditions. Anaerobic microorganisms can also degrade these polymers under anoxic conditions. The primary products then are CO2, H2O, CH4 and biomass under methanogenic condition or H2S, CO2 and H2O under sulfidogenic condition. The environmental conditions decide the group of microorganisms and the degradative pathway involved. Ultimate degradation of recalcitrant

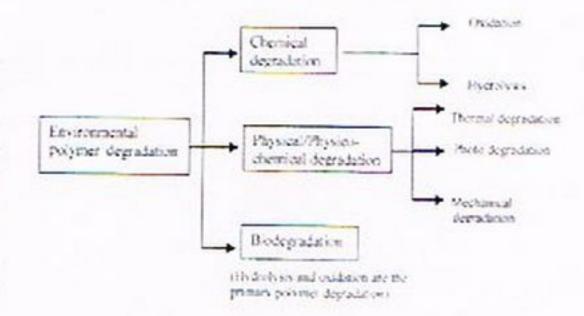


Fig. 1—Overview of degradation of polymers (Adapted from Vasile).

synthetic polymers may take several hundred years²⁴⁻²⁸.

Additives, antioxidants and other stabilisers added to commercial polymers may be toxic to the organisms or may slow down the rate of biodegradation.

Strategies used to Characterize Biodegradability of Polymers

As mentioned before, the high molecular weight polymers are degraded first into oligomers, some of which might be water soluble and then they are further broken down into organic intermediates. The intermediate products may be acids, alcohols, ketones, etc. The following strategies are used to assess and monitor the biodegradation of the polymers:

- Accumulation of biomass (experimentally determine the growth rate of microorganisms with the polymer as the sole carbon source)
- Oxygen uptake rate
- Carbon dioxide evolution rate
- 4. Products of reaction using chemical analysis
- Surface changes
- Changes in the mechanical and physical properties of the polymer⁸

Analytical Techniques

Several analytical techniques have been used to monitor the extent and nature of biodegradation (Fig. 2). These characterisation techniques are meant to study the mechanical, chemical, and physical properties of the polymer before and after degradation, which will help in understanding the extent as well as the mechanism of degradation. The study of mechanical properties comprises measuring of the tensile strength, elongation at fail and modulus of the polymer by using Instron. The physical properties of the polymers monitored are: morphology (microcracks, embrittlement using SEM, transmission optical microscopy), density, contact angle, viscosity,

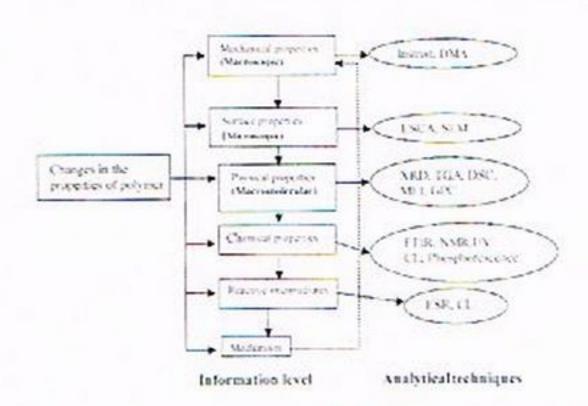


Fig. 2—Different levels of investigations on polymer biodegradation.

molecular weight distribution (using GPC), melting temperature (Tm), glass transition temperature (Tg) (doing TGA and DSC) and changes in the crystalline and amorphous regions (X-ray diffraction, SAXS and WAXS). The changes in the chemical properties that could be measured include formation or disappearance of functional groups as determined by FTIR. The molecular weight and molecular weight distribution of the degraded products or intermediates are characterised by techniques such as TLC, GC, GCMS, CL, MALDI-TOF, NMR (Fig. 3)5.23. The level of information derived from each technique, as shown in Fig. 2, increases as one moves downwards thereby understanding the mechanism of biodegradation. CO2 evolution is measured by using GC50, titrating with barium hydroxide41. Biofilm studies can be carried out using the acridine orange or BacLight bacterial viability kit57. The metabolic activity of the cells in the culture as well as in the biofilm can be done by ATP assays22, protein analysis and FDA analysis28. Thermally stimulated current spectra obtained from electret-thermal analysis reveals the electric polarization properties of polymer which is used for investigating biodegradation. Corona discharge pretreatment of polymers showed better results compared to UV treatment 13.27.

Factors Affecting Biodegradability

Biodegradability of the polymer is essentially determined by the following important physical and chemical characteristics:

- Availability of functional groups that increases hydrophilicity
- Size, molecular weight and density of the polymer

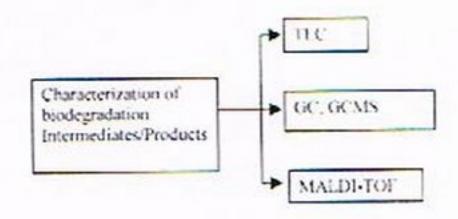


Fig. 3-Techniques used to characterize the degraded products.

- 3. Amount of crystalline and amorphous regions
- Structural complexity such as linearity or presence of branching in the polymer
- Presence of easily breakable bonds such as ester or amide bonds as against carbon-carbon bonds
- 6. Molecular composition (blend) and
- Nature and physical form of the polymer such as whether it is in the form of films, pellets, powder or fibres^{8,27}

Mechanism of Biodegradation of Polyolefins

In general, polyolefins are inert materials not susceptible to microbial attack because of the following reasons:

- Hydrophobic backbones consisting of long carbon chains that give high resistivity against hydrolysis³
- Addition of antioxidants and stabilisers during their manufacture which keeps polyolefins from atmospheric oxidation³
- 3. High molecular weight (from 10,000 to 40,000)
- 4. High packing density⁸

Even though PP is a polyolefin and prone to oxidative degradation similar to PE, the substitution of methyl in the place of hydrogen in the β position makes it more resistant to microbial attack, as already discussed in the factors affecting biodegradability (namely structural complexity)⁸.

The decreasing order of susceptibility of polymers to degradation in soil mixed with municipal refuse was PE>>>>LDPE>HDPE as revealed by analysing the weight loss of samples, CO₂ evolution, changes in tensile strength, changes in FTIR and bacterial activity in the soil¹².

Studies reported on biodegradation of PP are given in Table 1. As evident from the table, the work carried out in this area is scarce. Apart from fungal species (Aspergillus niger), microbial communities such as the species of Pseudomonas and Vibrio have been reported to biodegrade PP²³. A decrease in viscosity and

	Table 1—Variou	is literature report	s on biodegradation	of polypropylene	and its blends	
Title of the paper	Polymer	Organism	Conditions	Analytical techniques	Observation	Reference
Isotactic polypropylene biodegradation by microbial community	Isotactic polypropylene	Microaerophilic microbial community	Mineral medium containing sodiumlactate & glucose	IR, NMR, GC-MS	Organism & mycelia with known adaptability & metabolic flexibility can degrade isotactic PP	23
UV-Irradiated biodegradability of ethylene-propylene copolymers	Ethylene- propylene copolymers	Fungal species	Composting & culture environments	FTIR, SEM, VISCOSITY	Viscosity decrease & increase in carbonyl/hydroxyl region in FTIR	31
Biodegradation of γ- sterilised biomedical polyolefins	Isotactic polypropylene	Fungal species	Composting & culture environments	FTIR, SEM, VISCOSITY	Viscosity decrease & increase in chain scission	32
Blends						
Calorimetric & thermogravimetric studies of UV-irradiated polypropylene/starch- based materials aged in soil	Polypropylene/ starch	Soil	Soil burial tests	DSC and TGA	Biodegradation not affects the thermal stability, photooxidation decrease the thermal stability of the mixture	18
Effect of short wavelength UV- irradiation on ageing of polypropylene/cellulose compositions	polypropylene/ cellulose	Soil	Composted in garden soil	ATR-FTIR, TENSILE & SEM	Significant mechanical and surface changes found	20
Mechanical behavior of biodegradable polyoletins	(HDPE)/ polypropylene (PP)	Soil	Soil burial tests	DMM, VISCOELASTI C & DSC	A significant change in mechanical behaviour observed	36
Structure & properties of degradable polyolefin- starch blends	polyolefin- starch	Phanerochaete chrsosporium	Liquid fungus culture & soil burial test	Tensile DMTA, GPC, intrinsic viscosity, FTIR, & optical microscopy	Increased susceptibility to biodegradation	6
Enzymatic degradation of plastics containing polycuprolactone	PCL/PP	Rhizopus arrhizus lipase	Enzymatic condition	SEM & SPECTRO- METRIC	Blends of PCL and LDPE or PP retained high biodegradability of PCL	33
Thermal degradation of polypropylene/starch based materials with enhanced biodegradation	Polypropylene/ starch based materials	Soil	Soil burial tests	TGA, FTIR	Biodegradability observed more in starch based material rather than PP matters	15
Characterization by thermal analysis of HDPE/PP blends with enhanced biodegradation	Blends of HDPE/PP with different biodegradable additives	Soil	Soil burial tests	TG, DSC and dynamic- mechanical spectroscopy	Additive more affected by degradation than the polymeric matrix. Changes both in the crystalline morphology and activation energies of relaxation processes happens at different time & depends on the additives used	52

formation of new groups namely carbonyl and hydroxyl were observed during the degradation process32,33. Except for one report23, all the studies deal with degradation of pretreated PP. The pretreatment techniques reported range from UV-irradiation 17,20,32, ysterilization33 treatment14. and thermal These pretreatments either decrease the hydrophobicity of the polymer thereby making it more compatible with the organism or introduces groups such as C=O or -OH, which are more prone to degradation. It is reported that UV-treated PP sample is more susceptible to degradation than LDPE32. Biodegradation polypropylene/starch or polypropylene/cellulose blends has been reported using soil organisms. It is observed that the organisms easily degrade starch or cellulose leaving behind the polymer. These carbohydrates or fillers also increase the adhesion of the organisms to the surface of the polymer 5-20. Polycaprolactone (PCL) blended PP has also been reported to degrade in the presence of lipase31. PCL is an ester and since lipase is well known to degrade ester linkages, degradation of this polymer is facile. Lipase cannot affect the carboncarbon present in PP. There are no reports available on the effect of tacticity on the nature and rates of biodegradation as well as on the use of marine organisms to achieve biodegradation.

Biodegradation of isotactic polypropylene without any pretreatment is reported with one of the community designated as 3S among the four microbial communities (designated as 1S, 2S, 3S and 6S) adapted to grow on starch containing polyethylene obtained from enrichment culture. Pseudomonas chlororaphis, P. stutzeri, and Vibrio species were identified in the community 3S. TLC, GC-MS, FTIR and NMR analysis of dichloro methane extracted products confirmed the mixtures of different hydrocarbons with degrees functionalisation along with aromatic esters, which are added to the PP as a plasticiser. Sodium lactate and glucose had a co-metabolic effect. Starch enhances the adhesion of the microorganisms and also acts as a co-metabolite23.

The degradability of PCL blends such as PCL with polystyrene (PS), poly-ethylenetelephthalate (PET), and polyhydroxybutyrate (PHB) were less when compared to the degradability of PCL blended with LDPE or PP. This was due to the miscibility of PCL with conventional plastics such as polyolefins. High biodegradability of PCL was observed with PCL-LDPE and PCL-PP blends³³.

Outdoor soil burial tests were done on the samples of a HDPE and PP blend with different biodegradable additives. DSC analysis of these polymers with different additives after a year showed no change in melting temperature and fraction of crystalline region. Therefore, it was concluded that the biodegradation begins at the amorphous region rather than at the crystalline region. Biodegraded HDPE/PP blends were more brittle in nature compared to non-degraded second

Mechanical, rheological and susceptibility for natural degradation of polymer starch blends mainly depends upon the content, properties of starch, kind and concentration of additives added with the plastics. LDPE demonstrated lower degradability as compared with polypropylene in the presence of epoxidised rubber. The biodegradation of polymer along with the starch phase was observed in few cases.

The biodegradability of the UV-irradiated films of isotactic polypropylene (i-PP), ethylene-propylene copolymer and LDPE was studied in composting and A. niger culture. Increase in the rate of carbonyl and hydroxyl groups, decrease in the intrinsic viscosity and increase in chain scission after UV-irradiation has been reported. Decrease in the carbonyl region in FTIR was confirmed by the utilization of oxidized polymers by the microorganisms. The copolymer EPF-30R (having 7.7% ethylene) degraded faster than EPQ-30R (having 15.1% ethylene) demonstrating the effect of the composition of copolymer on biodegradability. PP was found to be more susceptible to microbial attack than LDPE. Weight loss and surface erosion were also reported.31 Additives are more susceptible to degradation rather than the HDPE and PP in HDPE/PP blends in outdoor soil burial test. Changes in the crystalline morphologies and activation energies of the relaxation process were confirmed by thermal analysis⁵².

Accelerated photo- and bio-degradations were reported with PP/cellulose blends when compared with pure PP in garden soil compost²⁰. γ-Sterilization of PP, LDPE and E-P copolymers were reported to have the same kind of effects as mentioned for UV-irradiated films³². Colorimetric and thermogravimetric studies on photo-degradation of polypropylene and a starch biodegradable additive mixture showed decrease in the crystallinity content due to free radical assisted chain scission, followed by biodegradation in soil, which later increased crystallinity due to the break down of chains in the amorphous region of the starch¹⁸.

Studies carried out on polyethylene biodegradation have been mentioned in Table 2. Unlike

Polymer	Organical	A contract of the state of the	4734	Ph. P.
Torythan	Organism	Analytical techniques used	Observation	Referenc
LDPE	Fungi Aspergillus niger, Penicillum funtculosum, Paecilomyces variotti, & Chocledum virens Bactetia Streptomyces badius, S setnii & S. viridosporous	DSC, FTIR, GPC & SEM	Molecular weight reduction, increase in carbonyl double bond groups, erosion on the surface of polyethylene is due to the microorganism	1.1
LDPE/ starch	Arthrobucter paraffineus	Gas chromatography- mass spectrometry, X-ray diffraction, size exclusion chromatography, FTIR, UV-Vis spectroscopy, DSC and SEM	Decrease in value of crystallinity, microorganism consumes carboxylic acids (carbon) evidenced by gas- mass spectrometry product	5
LDPE/ starch blends	Soil microorganisms, sludge microorganisms	Tensile strength, elongation, weight loss & SEM	85% percentage of elongation and 50% weight loss in 6 months	7
Polyethylene		Tensile strength	Heat treatment 70°C for 10 d. samples showed 60% elongation reduction in Streptomyces sp & 46.5% in fungi	40
HDPE/PP/ blends	Soil microorganisms	DSC, viscoelastic properties	Under soil burial conditions HDPE/PP blends altered mechanical behaviours	36
LDPE blends	Soil organisms	DSC	48% increase in crystallinity index	8
LDPE	Fungus	Contact angle and FTIR	Formation of carbony groups by oxidative process	27
LDPE/12% starch blend LDPE	Fungus Phanerocheate chrysoportum	Viscosity, percentage of elongation, CO ₂ evolution, FT-IR	Molecular weight reduced from 90,000 to 50,000 in 6 months. FT-IR showed strong absorbance in the region 1650-1860 cm ⁻¹ , 56% percentage of elongation in 3 months, increases CO ₂ evolution after 45 d of incubation	41
LDPE	Penicilium pinophilium & Aspergilius niger		niger. Decreases crystallimity, crystalline lamellar thickness. Increases carbonyl index	42
	LDPE/starch LDPE/starch blends Polyethylene HDPE/PP/ blends LDPE LDPE LDPE	Penicillum funiculosum, Paecilomyces variotii. & Cliocledium virens Bacteria Streptomyces badius. N setnii & S. viridosporous LDPE/ starch LDPE/ Soil microorganisms, sludge microorganisms Polyethylene Fungi Mucor rousii & Bacterium Streptomyces spp. HDPE/PP/ blends LDPE blends Soil organisms LDPE Fungus LDPE Fungus LDPE/12% Fungus LDPE Fungus LDPE/12% Fungus Phonerocheate chrysoportum LDPE Penicilium pinophilium &	LDPE Fungi Aspergillus niger, Penteulium funculostum, Paecilomyces variotii, & Cliochidium virens Bacteria Steptomyces badius, N setni & S. viridosporous LDPE/ Arthrobacter paraifineus LDPE/ Soil microorganisms, sludge starch blends microorganisms starch microorganisms starch blends microorganisms Polyethylene Fungi Mucor trucii & Bacterium Streptomyces spp. HDPE/PP/ Soil microorganisms Bacterium Streptomyces spp. DSC, viscoelastic properties DSC, viscoelastic properties LDPE Fungus Phonesocheme chrysopartum LDPE Penteilium pinophilium & Aspergillus niger LDPE Penteilium pinophilium & SEM DSC, X-ray diffraction, FTIR & SEM	LDPE Fungi Aspergillus niger Penecilium funculorum, Pacellomyces various & SEM SEM SEM SEM SEM Section of the meroseganism virens Bacteria Streptomyces balais, 8 Semit & 8 virilosporous Semit & 9 virilos Semit &

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Polymer	Organism	Analytical techniques used	Observation	Reference
I.DPE/starch blends/starch phthalate	Soil organisms	Mechanical properties, DSC, melt flow index & SEM	Tensile strength & clongation at break increased in LDPE/starch phthalate blends compared to the LDPE/starch blends	11
LDPE	Fungus Penicillum simssimum	HT-GPC, FT-IR	HT-GPC & FT-IR results showed double bonds of PE cut by Fungus P. samestmann	44
Polyethylene/ wax	Microbial consortium	GPC	Weight loss was 31.5%	45
Polyethylene	Soil microorganisms	Bioassmilations of product was evaluated	60% bioassimilation after 180 d	37
			1 1000 00 111	10
LDPE pro- oxidant additives	Soil microorganisms	FTIR and SEM.	in 18 months, carbonyl & double bond relative intensities of the carbonyl bond at 1715 cm-1 & double bond at 1650 cm-1	38
Degradable polyethylene (EPI TAPA)	Bacteria Rhodococcus rhodochrous, Cladosporium cladosporoides, Nocardia asteroides	Epitlurocent microscopy, SEM, FT-IR	Increased absorbance of carbonyl groups & double bond formation in 6 months 60% immeralization produced in 6 months	
Polyethylene	Unidentified three white rot fungi & Fusarium redolens	CO ₂ Evolution	Increases 0.5% CO ₂ evolution in 2 years of incubation	45
HDPE	Fusarium redolens, Acremonium kiliense, Aspergillus vesicolor & Verticillium lecanii	CO ₂ Evolution	Mixed culture of organism showed more degradation compared to single pure - culture by estimation of CO ₁	.46
HDPELDPE and 9% starch polyethylene	Heterotrophic bacteria	Weight loss, tensile strength, carbon dioxide production & IR	Starch PE - 82.76% loss of tensile strength, HDPE 5.33% and LDPE 13.04%, 36% weight loss in starch blend PE	12
LDPE	P. pinophilam, A. niger, Gliocladium virens & P. chrysosporium	DSC, FTIR & SEM	Thermal treated PE samples decreasing melting point and relative crystallimity. Degradation products were carbonyl & double bonds groups	47
	Polymer LDPE/starch blends/starch phthalate LDPE Polyethylene/ wax Polyethylene LDPE pro- oxidant additives Degradable polyethylene (EPI TAPA) Polyethylene HDPE and 9% starch polyethylene	I.DPE/starch blends/starch phthalate I.DPE Pungus Pentcillum simstirum Polyethylene/ Microbial consortnum wax Polyethylene Soil microorganisms LDPE prooxidant additives Degradable polyethylene (EPI TAPA) cladosporoides, Nocardia asteroides Polyethylene Unidentified three white rot fungi & Fusarium redolens HDPE Fusarium redolens, Acremonium killense, Aspergillus vesicolor & Verticillium lecunii HDPELDPE and 9% starch polyethylene I.DPE P. pinophilam, A. niger, Gliocladium virens & P.	LDPE Fungus Penteritum HT GPC, FT-IR	LDPE bungus Penterilum sinasiarum HT-GPC, PT-IR hierarilum sinasiarum GPC Weight loss was 31-55 LDPE providant additives Bacteria Rindococcus polyethylene Bacteria Rindococcus polyethylene Unidentified three white rot fungi & Fusarium redolens Acremonium killense. Aspergillus veiteolina Remonia polyethylene Unidentified three white rot fungi & Fusarium redolens Acremonium killense. Aspergillus veiteolina Remonia polyethylene Verticilium lecanii P. p. pmophilum, A. niger, Gliocladium virens & P. chrysosporium Decreasing meliting point and polyethylene P. pmophilum, A. niger, Gliocladium virens & P. chrysosporium Decreasing meliting point and polyethylene P. pmophilum, A. niger, Gliocladium virens & P. chrysosporium Decreasing meliting point and polyethylene P. pmophilum, A. niger, Gliocladium virens & P. chrysosporium Decreasing meliting point and polyethylene P. pmophilum, A. niger, Gliocladium virens & P. chrysosporium Decreasing meliting point and polyethylene P. pmophilum, A. niger, Gliocladium virens & P. chrysosporium Decreasing meliting point and polyethylene P. pmophilum, A. niger, Gliocladium virens & P. chrysosporium Decreasing meliting point and polyethylene P. pmophilum, A. niger, Gliocladium virens & P. chrysosporium Decreasing meliting point and relative crystallinity. Degratation products were

	Table 2—Var	ious reports on biodegradation	of polyethylene and its	blends: Contd.:	
Title of the paper	Polymer	Organism	Analytical techniques used	Observation	Reference
Electret-thermal analysis to assess biodegradation of polymer composites	LDPE/starch	Bacteria Baccilus, Clostridium & micrococcus Fungi Aspergillus, Penicillum & Mucor	DSC, FTIR, SEM & Physico-Mechanical testing	Biological erosion of polyethylene by oxidative process	13
DSC, FTIR characterization of biodegradation of polyethylene	Polyethylene	Fungi A. niger	DSC & FTIR	Decreased amorphosity of the sample and relative intensity of carbonyl bond formation	48
Colonization, biofilm formation and biodegradation of polyethylene by a strain of Rhodococus rubber	LDPE blends	Rhodococus rubber	FTIR, SEM & weight loss	Carbonyl index reduced 66%, enrichment medium supplement with 2% mineral oil showed 50% degradation after 30 d incubation	28
Synergistic effect of combining UV sunlight- soil burial treatment on the biodegradation rate of LDPE/starch blends	blends	Soil organisms	DSC, FT-IR, tensile strength & SEM	Starch blend PE exposed UV radiation & soil burial samples showed 66% degradation	14
Biodegradation of polyethylene by the thermophilic bacterium Brevibacillus borstelensis	LDPE	Brevibacillus borstelen	DSC, FT-IR	31% Molecular weight reduction in 30 d	51
Study and development of LDPE/starch partially biodegradable compounds	LDPE/starch blends	Sludge microorganisms	Tensile strength & SEM	Reduction in tensile strength & elongation properties, LDPE degraded in the amorphous region responsible for oxidative process	16
Acquired biodegradability of polyethylene containing pro-oxidant additives	LDPE HDPE/blends	R. rhodochrous, N. asseroids, Aspergillus flavis, C. cladospoides	ATP, ADP assays, Size exclusion chromatography, Micr oscopy techniques & NMR	R. rhodochrous & N. astroides found to be most active for molecular weight reduction	22
Effect of compatibiliser on the biodegradation and mechanical properties of high content starch/low density polyethylene blends	LDPE/starch blends	Soil organisms	Mechanical properties, weight loss, melt flow index & SEM	65% weight loss increase in 14 d	17
Polyethylene biodegradation by developed <i>Penicillium-</i> <i>Bacillus</i> biofilm	Polyethylene	P. frequentans B. mycoides	Microscopy, weight loss, gas chromatography	Weight loss of preheated polyethylene treated with fungi showed 7.150% & without preheating treated with showed 6.657%	50
Photo biodegradation of low density polyethylene/banana starch films	LDPE/starch blends	Soil microorganisms	FTIR, tensile strength, clongation & weight loss	Increased carbonyl index & Tensile strength & elongation at break increased in LDPE/starch blends	19
Biodegradation potential of some barrier-coated boards in different soil environments	Polyethylene & Polyester	Soil microorganisms	DSC & FTIR	Under soil burial condition PE/Polyester blends affect mechanical behaviors	49
					Contd.:

	Table 2-Va	rious reports on biodegrad	lation of polyethylene and it	s blends: Contd.	
Title of the paper	Polymer	Organism	Analytical techniques used	Observation	Reference
Modification of polymers by protein hydrolysate-A way to biodegradable materials		Aspergillus oryzae	Mechanical strength properties	Polymer blend with 20% HP (Protein hydrolysate) shows 35% biodegradation with acceptable range of mechanical strength whereas polymer with 40% HP shows 50% biodegradation with processor mechanical strength properties.	

polypropylene, more research articles are published on studies relating to biodegradation of PE. Fungi that include A. niger, Penicillium funiculosum, Fusarium redolens, and A. vestcolor, and soil microorganisms (mixed culture as well as Rhodococcus rhodochrous, Cladosporium cladosporoides) have been reported to degrade neat PE 11.19.30.30. DSC or FTIR and other mechanical and physical techniques such as weight loss, changes in tensile strength have been the commonly used analytical techniques to monitor the nature of biodegradation. Thermal, UV, photo and corona treated PE has been found to degrade faster than the untreated polymer. Biodegradation of starch blended and modified PE with protein hydrolysate has also been studied 10.

Photooxidation is the triggering step in the oxidative degradation of polyethylene. UV radiation leads to radical formation, followed by the absorption of oxygen resulting in end products with carbonyl groups. Additional UV exposure causes the carbonyl group to undergo Norrish type I and/or Norrish type II degradation which leads to the cleavage of C-C bond and thus leading to the formation of oxidised low fragments. weight molecular photooxidation leads to the formation of low molecular weight fragments and thus increases the hydrophilicity of the polymer (4,18-212828-28-31 photooxidation mechanism shown in Fig. 4. comprises both the formation of carbonyl group as well as Norrish type I and type II. Thus, photooxidation enhances the susceptibility of the polymer to nucrobes. The resulting carboxylic acid from the photooxidation and w-oxidation of long chain hydrocarbons (similar to the biotic degradation of paraifin-C_{10,20}) enters the β-oxidation pathway as shown in Fig. 4. Later, the two carbon acetyl CoA. enters the TCA cycle and gets completely converted into carbon dioxide and water 5.26,27,40,31

Cell homogenates from P. putida and Bucillus brevis were found to degrade PE films by oxidative

degradation resulting in the formation of terminal hydroxyl, ketone and ester groups. The presence of alcohol dehydrogenase was confirmed indirectly in the degradation reaction by inhibition studies. The known lignin degrading bacteria S. virdosporus T7A. S. badnes 252, and S. setonni 75vi2 and the fungus Phanerochaete chrysosporum were used to assess their ability to degrade biodegradable polyethylene (polyethylene with 6% starch and pro-oxidant). The authors observed the accelerated pro-oxidant activity by heat treatment and UV treatment with different time period. Reduction in polydispersity and tensile strength were observed in biodegradable PE with bacterial treatment and not with the fungus. The veratryl alcohol lignin peroxidase activity was confirmed".

A. niger has been reported to degrade commercially available PE DSC analysis showed reduction in the amorphous region of the polymer". Biodegradation of LDPE was enhanced with Tween 80 in the presence of P. aeuroginasa. This study explains the role of nonionic surfactant in biofilm formation, as explained before it is a prerequisite for biodegradation process*6. Biodegradation of thermally oxidized LDPE with fungal cultures of A niger, Pencillium funicalasum, Paccilomyces various and Gliocladium virens was marked by the gradual decrease in carbonyl region (1715 cm) in FTIR Disposable polyethylene bags with 6% starch were subjected to biodegradation for a period of four weeks by eight different species of Streptomyces and the fungi, Mucor rouxii and A. flavus. Weight gain was seen after degradation with few Streptomyces species, whereas a slight loss of weight was observed with S. aburaviensis, S. parvullus, S. nigellus and A. Jlavus. Reduction in percentage elongation with Streptomyces and fungal cultures were 28.5% and 46.5%, respectively. Thermally, treated film incubated with Mucor had 60% reduction in tensile strength.

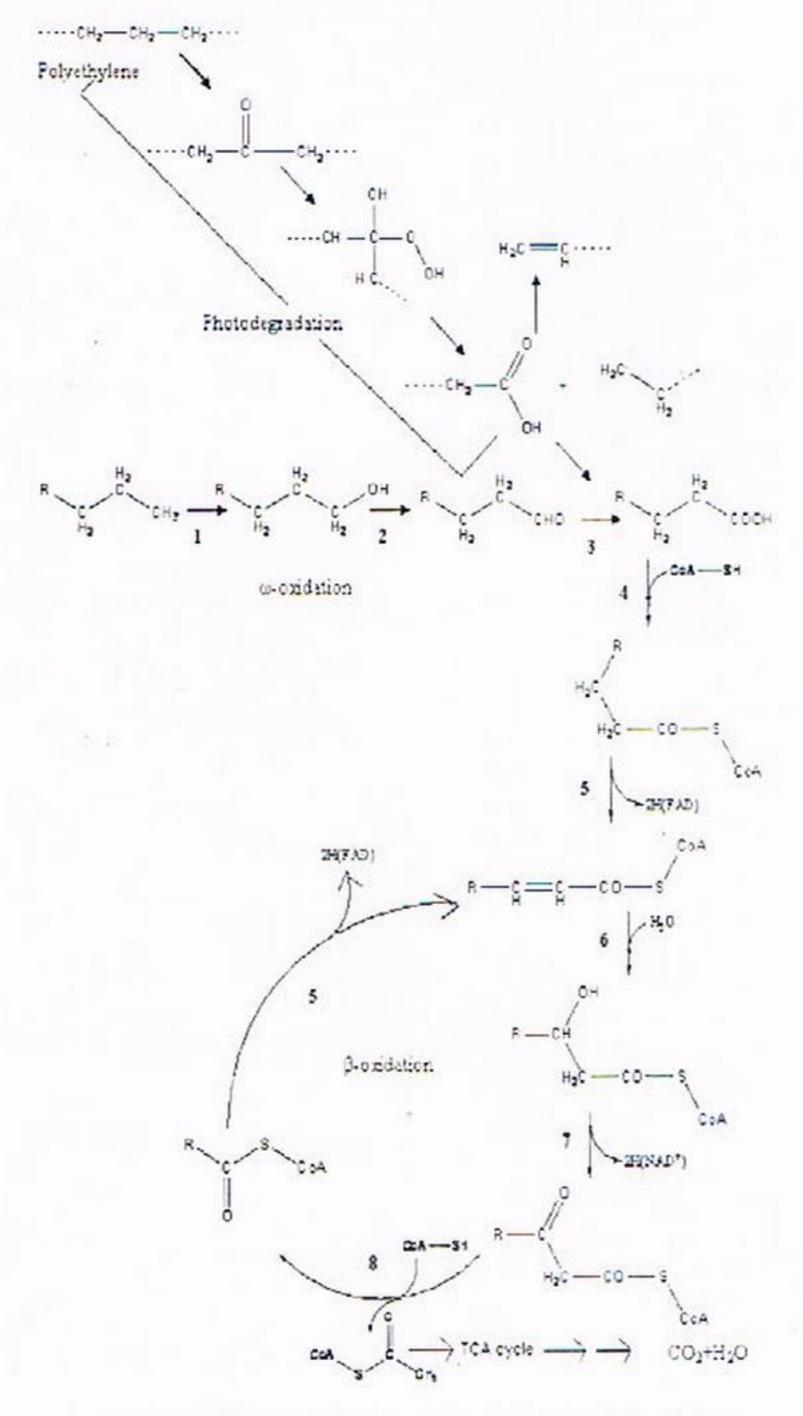


Fig. 4—Mechanism of biodegradation of polyethylene (Adapted from Vasile).

The bacteria, Arthobacter parraffineus was found to degrade LDPE in three years by utilising carboxylic acid formed during thermal oxidation. The utilization was through the β-oxidation mechanism that yields the degradation products like acetyl coA and propionyl CoA. 3- methyl-3-octanol and 1-hexadecanol were detected in biotic environment with series of n-alkanes such as C21-26. These were microbiologically metabolised by the oxidation of carboxylic acid through β-oxidation.

Rate of degradation of octonated starch is slower than pure starch. OCST-LDPE blend and octonated starch was subjected for six month soil burial test, which showed weight loss and reduction in mechanical properties. SEM analysis of OCST-LDPE blend showed the presence of holes on the surface, which confirmed the degradation of OCST region in blend⁷.

Corona discharge treatment was found to be more effective towards colonisation of microorganisms on food packaging grade LDPE films with little effect on the mechanical properties as compared to UV treatment. This suggests that corona discharge treatment is affecting the hydrophobicity of the surface of the polymer and not penetrating it. A reduction in hydrophobicity of the LDPE from (92° to 66.6°) was also reported27. The pH of Phanerochaete chrysosporium inoculated soil with polyethylene decreased at a faster rate. Biomass, biological activity and CO2 evolution was higher in inoculated soil. Analysis of the mechanical properties showed that decrease in the percentage elongation is faster in the inoculated soil compared to the uninoculated soil. Viscosity analysis of the polymers with regular intervals also showed the same trend41. A similar study was performed by Yamada-Onodera et al using Triton X-100. Improvement was observed in the growth of Penicillium simplicissimum YK, however, there was no utilisation of Triton X-100. FTIR analysis confirmed the utilisation of polyethylene by the fungus".

Thermal treatment of LDPE-TDPA (Pro-oxidant additive) in aerobic conditions showed substantial polymer fragmentation with loss of mechanical properties in 11 d. 26% of biodegradable and solvent extractable fraction was obtained after thermal oxidation for 20 d. 50-60% carbon dioxide evolution was observed in 18 months of further treatment with soil microorganisms³⁸. Temperature is the crucial factor in determining the rate of thermo-oxidation

whereas the effect of concentration of oxygen on the rate of thermo-oxidation is insignificant37. A. niger, G. virens, Penicillium pinophilum, Phanerochaete chrysosporium showed biodegradation on thermally treated or accelerated ageing treated (AAT) LDPE in 9 months. The biodegradation was evaluated by observing decrease in the onset of melting temperature (To) and melting temperature Tm and relative crystallinity. Highest mineralization (3.26%) values were obtained with AAT. Superficial growth of microorganisms occurred and penetration of hyphae was observed in the oxidised sample". Synergistic effect of combining UV treatment and soil burial test was reported by Abd El-Rehim et al14. Electret-thermal analysis used in the electric polarization of dielectrics was used to investigate biodegradation of LDPE- starch blended polymer in 6 months. These studies were based on the assumption that biodegradation process of polymer material can cause transformation in their electrically nonequilibrium structure. Thermally stimulated current spectra (TSC) of PE films exposed to various ageing conditions in soil were reported. After ageing, new peaks were detected on spectra. FTIR results showed formation of functional groups. Reduction in melting was reported in DSC analysis. The degree of biological damage of the films was a function of starch content of the composites. The predominant microbial taxa in composites were Bacillus, Clostridium, Micrococcus, Aspergillus, Penicillum and Mucor 13.

Rhodococcus ruber C208 was isolated from the surface of the PE in polyethylene waste burial site by two step culture-enrichment protocol. Weight loss of 8% of photo-oxidised PE was observed in four weeks. This is higher than the rates already reported (3.5% to 8.4% after 10 years)30. In contrast to Albertsson's report, increase in the terminal double bond after photooxidation was observed. This could be explained by Norrish type I degradation of the carbonyl residues. They have reported that the double bonds were observed after the biodegradation of short PE oligomers produced during photooxidation. The analysis of extracellular polysaccharides in the biofilm of C208 was 2.5 folds higher than protein, suggesting its role in biofilm formation. Biofilm showed higher viability even after 60 d of incubation. Cell surface hydrophobicity of R. ruber was studied by SAT (salt aggregation test) and BATH (bacterial adhesion to hydrocarbon) tests. Addition of mineral

oil to this culture enhanced the degradation of the PE film by about 50% after four weeks of incubation. SEM photomicrographs of the bacterial biofilm showed some localized degradation of the PE around the bacteria. Protein assay and FDA hydrolysis by extracellular esterases showed increase in the biofilm formation for the first 2 d of assays followed by a sharp decrease in biomass density. The authors have hypothesised a low cell population with a low growth rate consisting of cells that are able to utilise PE as a carbon source^{28,57}.

Brevibacillus thermophillic borstelensis, bacterium, was found to degrade polyethylene better than R. rubber, although the biofilm forming capacity of the former was not found to be as good as of the latter. Still it was able to show reduction in mass and molecular weight by 11 and 30%, respectively for UV irradiated polyethylene31. The LDPE and HDPE films after photo-oxidation and thermal oxidation corresponding to three years of outdoor weathering were incubated with R. rhodochrous and Nocardia asteroids. ATP assay was done to see the metabolic activity of the cells in culture and those adhered to the surface of the polymer. There was fast growth of microorganisms in the initial phase due to the availability of the low molecular weight oxidised products, which was followed by stabile metabolic activity. This was maintained for several months by the organisms utilising the polymer. The NMR analysis of the photo- and thermo-oxidized LDPE/HDPE aqueous extract revealed the presence of ethanol and formate, which are the end products of PE oxidation. This evidence supports the initial fast growth of microorganisms observed by ATP analysis. Nocardia formed dense filamentous mycelium on the surface. The size exclusion chromatographic analysis of the LDPE/HDPE after biotic and abiotic treatment showed no change in the molecular weight distribution indicating that the microbial attack was only on the surface of the polymer. The degradation due to both biotic and abiotic factors depended on the thickness of the polymer22.

Studies on biofilm formation by Penicillium frequentans and Bacillus mycoides showed that P. frequentans formed a network of mycelia on degradable polyethylene (DPE-chemical or photoinitiator added polyethylene), which was colonised by B. mycoides. The biofilm formation increased the biodegradability of P. frequentans by 14 folds. In general, homologous gene has been found in

the genome of some Bacillus species that produce alkane monooxygenase. The degradation was checked with weight loss, microscopic studies to visualise biofilm formation and CO₂ production using GC⁵⁰.

Conclusions

literature This review discusses the biodegradation of PE and PP. Most of the examples deal with fungi and bacterial based degradation. Pretreated polymers degrade more easily than the untreated polymers. Also, degradation is more facile with starch and cellulose blended polymers. Cell surface hydrophobicity and addition of surfactants showed an important role in biofilm formation, which is prerequisite for biodegradation. Degradation leads to decrease in molecular weight, tensile strength and viscosity, formation of new functional groups such as carbonyl, hydroxyl, etc. Based on the literature one could conclude that in order to enhance biodegradation of PP or PE the following approaches could be adopted:

- Modify the polymer for microbial utility by the (i)Addition of natural polymers and/or prooxidants to PP; (ii) Modification of polymers by protein hydrolysates; and (iii) Pretreatment of the polymer.
- Modify the microbes to utilise the polymer by

 Modifying medium composition, and thus
 enhancing the utilisation of polymer; and (ii)
 genetically modify the microorganism to
 utilise the polymer.
- III. Overexpress the enzyme, which is responsible for degradation and purify it and utilise for this purpose. Strategies II and III require the understanding of mechanism of microbial degradation of these polymers.

APPENDIX - ABBREIVIATIONS

AAT	 Accelerated ageing treatment
ATP	-Adenosine TriPhosphate

Fourier Transform Infrared Spectroscopy

CL -Chemilluminesence

DPE

DMA -Dynamic Mechanical Analysis

DSC -Differential Scanning Calorimetry

ESCA -Electron Spectroscopy for Chemical

-Degradable Polyehtylene

Analysis

ESR	-Electron Spin Resonance					
FDA	-Fluorescien DiAcetate					
FTIR	-Fourier Transform Infrared					
	Spectroscopy					
GC-MS	-Gas Chromatography - Mass Spectrometry					
HDPE -	-High Density Polyethylene					
HTGPC	-High temperature gel permeation chromatography					
i-PP	-Isotactic Polypropylene					
LDPE	-Low Density Polyethylene					
	TOF-Matrix Assisted Laser					
	Desorption/Ionisation - Time of flight					
MFI	-Melt Flow Index					
NMR	-Nuclear Magnetic Resonance					
	Spectroscopy					
NY	-Nylon					
OCST	-Octonated starch					
PCL	-Polycaprolactone					
PET	-Polyethylenetelephthalate					
PHB	-Polyhydroxy butyrate					
PS	-Polystyrene					
SAXS	-Small Angle X-ray Scattering					
SEM	-Scanning Electron Microscopy					
Tg	-Glass Transition temperature					
TGA	-Thermogravimetric analysis					
TLC	-Thin Layer Chromatography					
Tm	- Melting temperature					
TSC	-Thermally Stimulated Current					
	Spectra					
UV	-Ultra Violet Spectroscopy					
WAXS	-Wide Angle X-ray Scattering					
XPS	-X-ray Photoelectron Spectroscopy					
XRD	-X-Ray Diffraction					

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